

Effects of Adding TiO₂ Nanoparticles on Flexural Strength and Hardness of Two New Commercial Flowable Dental Composites

Fatemeh Koohepeima¹, Mohammad Javad Mokhtari², Sajjad Khaksabz¹

¹Department of Operative Dentistry, Biomaterial Research Center, School of Dentistry, Shiraz University of Medical Sciences, Shiraz, Iran

²Department of Biology, Zarghan Branch, Islamic Azad University, Zarghan, Iran

Received 14 July 2021 and Accepted 21 May 2022

Abstract

Introduction The present study aimed to investigate the effect of adding TiO₂ nanoparticles on the flexural strength and hardness of two dental composites. **Methods:** TiO₂ nanoparticles were prepared and added to two flowable dental composites (Beautifil Flow Plus and Clearfil AP-X Flow) at 0, 0.2, 0.5, and 1% (w/w) which was confirmed by SEM and TEM analysis. Mixing was manually performed using Lentulo Spiral Paste Carrier. The specimens were divided into 8 groups of 10 according to the type of composites and different concentrations of TiO₂ nanoparticles. Bar-shaped specimens (2×2×25 mm) were fabricated in a half-split stainless steel mold and cured for 40 s by an LED curing system. Flexural strength was evaluated using a universal testing machine. Surface microhardness was also measured by the Vickers microhardness tester. **Results:** For the two tested composites, flexural strength increased by 0.5% TiO₂ nanoparticles concentration. The flexural strength of Clearfil AP-X Flow combined with 0.2, 0.5 and 1% TiO₂ nanoparticles were 112.54±12.87 ($P>0.05$), 114.62±8.14 ($P>0.05$), and 99.92±6.23 ($P<0.05$), respectively. Also, for Beautifil Flow combined with 0.2, 0.5 and 1% TiO₂ nanoparticles were, 92.21±4.26 ($P<0.05$), 94.05±5.36 ($P<0.05$) and 74.17±9.43 ($P<0.05$), respectively. **Conclusion:** Adding TiO₂ nanoparticles to Clearfil AP-X Flow composite decreased the hardness; however, adding TiO₂ nanoparticles by 1% concentration increased the surface hardness of Beautifil Flow and reached its maximum value. Also, TiO₂ nanoparticles at very low concentrations enhance the flexural strength of dental flowable composites.

Keywords: Flowable Composite, Nanoparticles, Surface Hardness, Titanium Dioxide

Koohepeima F, Mokhtari MJ, Khaksabz S. Effects of Adding TiO₂ Nanoparticles on Flexural Strength and Hardness of Two New Commercial Flowable Dental Composites. *J Dent Mater Tech.* 2022; 11(2): 68-74.

Introduction

The consistency of most dental composite resins is desirable for clinical conditions; however, some cases require a type of composite resin with lower viscosity for better adaptability with the cavity walls. Therefore, a new category of “flowable composite resins” was introduced in late 1996. These flowable composites were created using the same size of small particles by reducing filler contents to about 40-50% (volume percent) compared to 55%-75% of conventional hybrid composites which can modify the viscosity of these materials (2, 3). Despite the fact that these types of dental composites provide a wide range of various benefits to practitioners, our literature review revealed a paucity in studies evaluating the characteristics and application techniques of these new materials (1, 4, 5). Several authors have suggested that the three-point flexural strength test is better than other test methods due to less complexity (6).

Some previous studies have focused on the pretreatment of inorganic monomers and fillers to improve the properties of flowable resin composites (7). These research studies have shown that fibers and microparticles may also be used to reinforce dental resin composites. It has been shown that combining small amounts of networked and/or short fibers into the composites can lead to increased strength (8, 9).

Using nanoparticles in dentistry has become a significant area of research (10, 11). Among these inorganic fillers,

TiO₂ has a highly efficient photocatalytic effect and is chemically stable (9).

Previous studies using TiO₂ nanoparticles in epoxy resins revealed that they can enhance toughness, stiffness, and strength (12). TiO₂ nanoparticles have many promising properties as an inorganic additive in resin composites. The nanoparticles were combined with cold-curing resin composites to achieve opalescence (13). In recent studies, dental resin composites and glass-ionomer cement with TiO₂ nanoparticles demonstrated superior mechanical properties (14). Some nanoparticles have been applied in attempt to add antimicrobial properties to dental materials (11). TiO₂ nanoparticles have been used as additives to many biomaterials due to their antimicrobial properties (15). Interactions between nanoparticles and cells have been extensively studied owing to their nanoscale size. The cytotoxicity of different nanoparticles and dental materials that have been investigated in recent studies revealed different results (16, 17).

Therefore, the present study aimed to evaluate the effect of TiO₂ nanoparticles on the flexural strength and hardness of two commercial dental flowable composites.

Materials and Methods

Preparation and characterization of TiO₂ nanoparticles

Titanium isopropoxide 97% (Sigma Aldrich, Saint Louis, MO, USA) was diluted in isopropyl alcohol (Merck) as the initial solution. Precipitation of TiO₂ nanoparticles occurred by adding alkaline (pH 8) distilled water. Distilled water was then used for washing the precipitate. Then the as-prepared precipitate was centrifuged and heated. Morphological analysis and measuring the size of the TiO₂ nanoparticles were performed using scanning electron microscopy (SEM) (XL30, Philips, Netherlands) and transmission electron microscopy (TEM) (JEOL, JEM-1011; Japan), respectively. X-ray diffraction (XRD) was used as a technique to determine the structure by analyzing the diffraction peaks (18). XRD pattern of the TiO₂ nanoparticles was obtained by a DX-1000 X-ray powder diffractometer (Dandong Fangyuan, Dandong, China).

Specimen preparation

TiO₂ nanoparticles were added to the Beautiful Flow (Shofu, Kyoto, Japan) and Clearfil AP-X Flow (Kuraray, Kyoto, Japan) composites. Initially, mixing was manually performed by spatulating TiO₂ nanoparticles in the flowable composite and then using a Lentulo Spiral Paste Carrier (19, 20). The Lentulo Spiral was inserted

into the mixture of composite resin and nanoparticles and poured into a pre-darkened syringe tube. After mixing for 15 min, the drill/grinder was extracted with attached Lentulo; a syringe plunger was inserted into the syringe tube and 80 bar-shaped specimens (2×2×25 mm) were fabricated in a half-split stainless steel mold (ISO 4049) and then cured by LED (Monitex, GT1200) with a power of 800 MW/Cm². The light guide was placed at the center of the specimen and against the glass slide and then the specimen was irradiated for a polymerization time of 20 s (as recommended by the manufacturer). After photoactivating the center of the specimen, the light guide was placed at the other section next to the center that overlaps the previous section; the same procedure applies to the other side of the center; both sides were irradiated as recommended by the manufacturer. Finally, the specimens were kept in distilled water at 37°C for 24 h before testing.

The specimens were divided into 8 groups of 10 according to the type of composites and different concentrations of TiO₂ nanoparticles (0%, 0.2%, 0.5%, and 1% w/w).

Flexural strength

The specimens were subjected to a three-point-bend test using a universal testing machine (Zwick Roell, Z020) at a crosshead speed of 0.5 mm/min to measure flexural strength. Flexural strength (in MPa) is determined using the following formula:

$$\text{Flexural Strength} = \frac{3FL}{2BH^2}$$

Where F is the failure load (Newtons), B is the specimen width (mm), L is the distance between the supports (mm), and H is the specimen height (mm).

Hardness

Ten samples of the respective composites were placed inside a stainless steel mold (8 mm in diameter and 2 mm thickness) for measuring the Vicker's microhardness of the composites, while a dark opaque paper background covered with a polyester matrix strip was used. This arrangement reduced the possibility of obtaining higher artificial hardness in that area. The mold was filled with the flowable resin composite and another polyester matrix strip was adjusted on the surface of the filled mold. A glass slide was pressed against the top layer of the polyester film to form a flat surface of composites and to extrude the excess composite resin. The composite material was then light-cured from the top for 20 s by an LED (Monitex, GT1200) with a power of 800 MW/Cm². The microhardness was calculated by applying a load of 200 g for 15 s using a microindenter (MHV-1000Z,

China), and the indentation size was recorded 15 s later. A total of 5 square indentations were produced in each specimen, and this process was repeated 3 times. Finally, the average indentation size was used to determine the Vickers hardness number (VHN) using the following formula (21).

$$HV = \frac{1854 \times F(g)}{D^2(\mu m)}$$

F is the load (g), HV is the microhardness value, and D is the dimension of the indentation (mm).

Statistical analysis

Data analysis was performed using SPSS software (version 16). The results of the study were analyzed using

one-way ANOVA test. A p-value ≤ 0.05 was considered statistically significant.

Results

SEM, TEM, and XRD analysis

SEM and TEM (Figure 1) were used to investigate the size and morphology of particles. The SEM and TEM images indicate that the TiO₂ particles are spherical with an average diameter of 30 nm. A typical XRD pattern of the as-prepared TiO₂ nanoparticles is shown in Figure 2. XRD pattern of pure TiO₂ shows five primary peaks at 27, 36, 39, 41, 44, 54, 57, 63, 64, 69, and 70, which can be attributed to different diffraction planes.

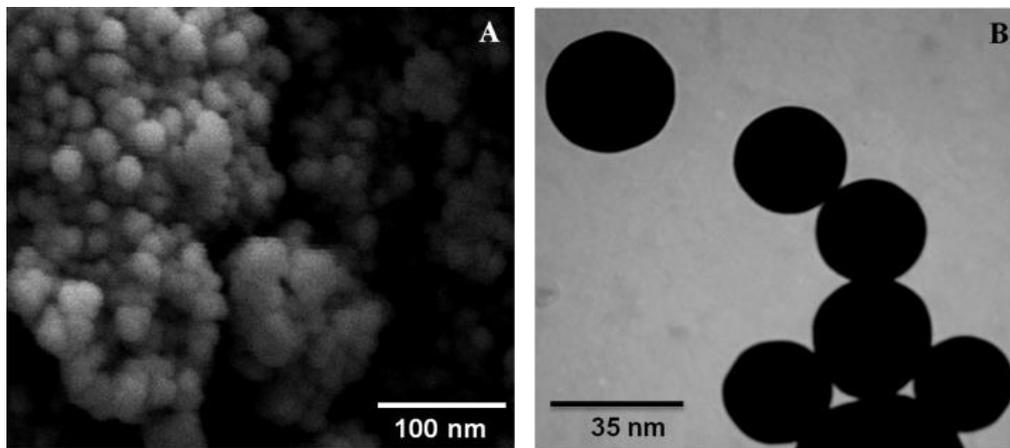


Figure 1. (A) SEM photography of TiO₂ nanoparticles, (B) TEM photography of TiO₂ nanoparticles

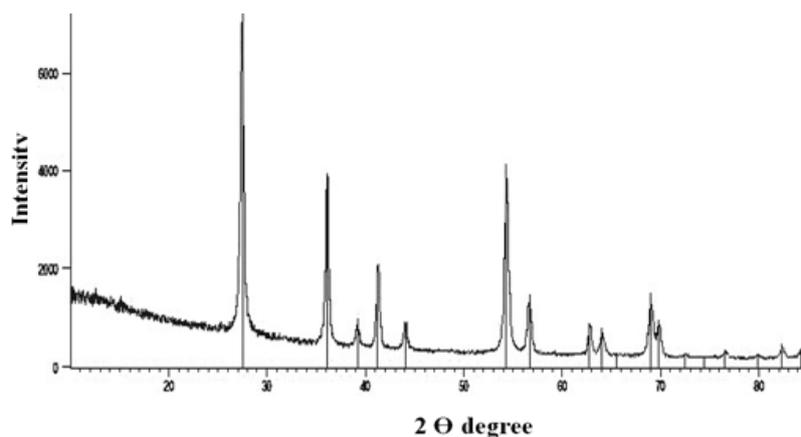


Figure 2. XRD report of TiO₂ nanoparticles

Scanning electron microscope

Figure 3 indicates the distribution of a 0.5% concentration of TiO₂ nanoparticles in composite resins.

SEM indicated that TiO₂ nanoparticles had a homogeneous distribution in Clearfil AP-X Flow and Beautifil Flow.

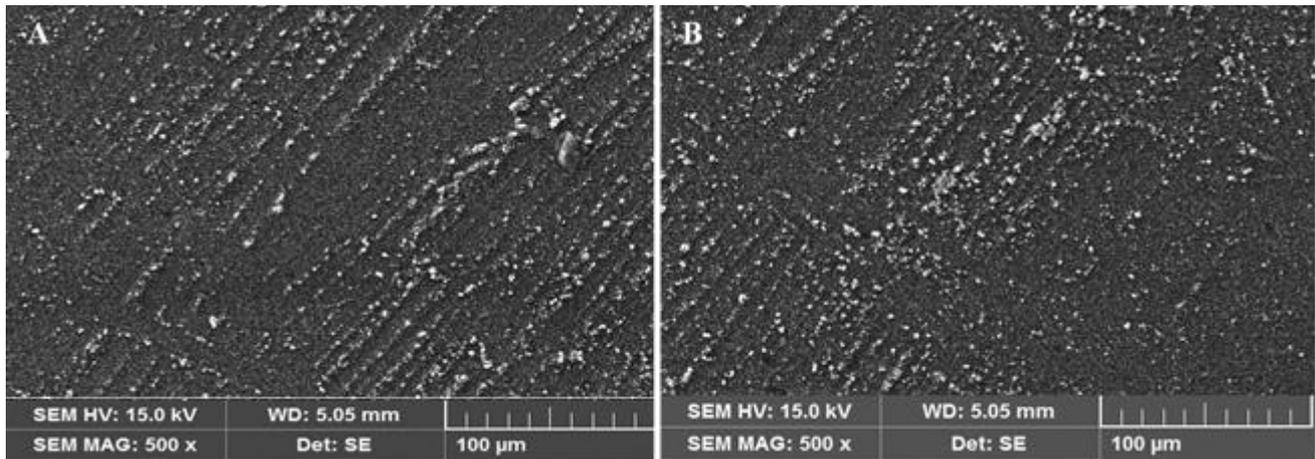


Figure 3. SEM images taken from cross-section of (A) Beautifil Flow with 0.5% TiO₂ nanoparticles and (B) Clearfil AP-X Flow with 0.5% TiO₂ nanoparticles 0.5 wt%

Flexural strength and Hardness

ANOVA test showed a significant difference in flexural strength of Clearfil AP-X Flow with 1% TiO₂ nanoparticles and Beautifil Flow with 0.2, 0.5, and 1% TiO₂ nanoparticles. The flexural strength of Clearfil AP-X Flow with 0.2, 0.5, and 1% TiO₂ nanoparticles were 112.54±12.87 ($P>0.05$), 114.62±8.14 ($P>0.05$), and 99.92±6.23 ($P<0.05$), respectively. Also, the flexural strength of Beautifil Flow combined with 0.2, 0.5, and 1% TiO₂ nanoparticles were 92.21±4.26 ($P<0.05$), 94.05±5.36 ($P<0.01$), and 74.17±9.43 ($P<0.05$), respectively. Flexural strength increased by 0.5% TiO₂ nanoparticles concentration and decreased by 1% for the two tested composites. The increase of flexural strength in Clearfil AP-X Flow is lower than that of the Beautifil Flow.

ANOVA test revealed a significant difference in hardness of Clearfil AP-X Flow with 0.5 and 1% TiO₂ nanoparticles and Beautifil Flow combined with 1% nanoparticles. The hardness of Clearfil AP-X Flow with 0.2, 0.5, and 1% TiO₂ nanoparticles were 86.22±1.69 ($P>0.05$), 78.08±3.02 ($P<0.05$), and 60.80±3.27 ($P<0.01$), respectively. Also, for Beautifil Flow combined with 0.2, 0.5 and 1% TiO₂ nanoparticles were 38.35±0.75 ($P>0.05$), 39.45±0.93 ($P>0.05$) and 41.41±3.24 ($P<0.05$), respectively. Results clearly indicated that the addition of TiO₂ nanoparticles to Clearfil AP-X Flow decreased the hardness of specimens. Adding TiO₂ nanoparticles with the concentration of 1% w/w to Beautifil Flow composites, increased surface hardness to its maximum value. The average values of flexural strength and hardness at different TiO₂ concentrations are displayed in Table I.

Table I. Mean flexural strength, standard deviations of each group (MPa), mean hardness, and standard deviations of each group

		TiO ₂ concentration				Significance
		0% w/w	0.2% w/w	0.5% w/w	1% w/w	
		(A)	(B)	(C)	(D)	**P<0.01
Flexural Strength	Clearfil AP-X Flow	111.12±5.86	112.54±12.87	114.62±8.14	99.92±6.23	A,D* B,D*C,D**
	Beautiful Flow	81.62±4.82	92.21±4.26	94.05±5.36	74.17±9.43	A,B* A,C** A,D* B,D** C,D**
Hardness	Clearfil AP-X Flow	89.26±4.30	86.22±1.69	78.08±3.02	60.80±3.27	A,C* A,D** B,C* B,D**
	Beautiful Flow	37.05±2.97	38.35±0.75	39.45±0.93	41.41±3.24	A,D* B,D*

Discussion

SEM and TEM images indicate that TiO₂ particles are spherical with an average diameter of 30 nm. In the present study, TiO₂ nanoparticles had a homogeneous distribution in Clearfil AP-X Flow and Beautiful Flow composites. Flexural strength increased by 0.5% TiO₂ nanoparticles concentration for the two tested composites. Results revealed that adding TiO₂ nanoparticles to Clearfil AP-X Flow decreased the hardness of specimens. For Beautiful Flow, adding TiO₂ nanoparticles at a concentration of 1% w/w increased surface hardness to its maximum value.

Scientists have conducted extensive research in aims of improving inorganic fillers in different types of dental resin composites. The current tendency is to reduce the size of filler particles from micrometer to nanometer scales (22). Using nanoscale fillers in resin composites seems to affect their biological and mechanical properties (23). TiO₂ nanoparticles are optimal for application in dentistry due to their high biocompatibility and satisfactory color (24). TiO₂ is considered a chemically stable and inexpensive nanoparticle that has been used for a wide range of environmental applications (15).

Clearfil AP-X Flow is a flowable and restorative composite resin. The inorganic filler particles of this composite are treated with a novel technology of surface coating. This treatment leads to a reinforced resin matrix

with a high filler loading of 81 wt% which is similar to universal composite resins. Beautiful Flow is another flowable restorative composite that consists of filler content of about 50 wt%. This lower filler content may result in less optimal mechanical properties. Bending the bar specimen occurred when the load was applied under the flexural condition.

Compressive stresses are mostly directed at the top surface of the sample; whereas tensile forces mainly affect the more inferior surfaces. The main advantage of flexural modulus for polymeric composites is that it describes the combined effects of tensile and compressive deformation (15). Based on the obtained flexural strength values, increasing the concentration of TiO₂ nanoparticles up to 0.5% w/w increased the flexural strength while adding a further amount of TiO₂ nanoparticles decreased flexural strength. Another important reason is a phenomenon called agglomeration in which fine nanoparticles accumulate around each other or other particles and create nanoclusters that weaken mechanical properties. An increase in TiO₂ content, causes agglomeration of these particles. Some structural defects such as micro-pores and micro-cracks may develop due to the agglomeration of TiO₂ nanoparticles at high percentages. Shirkavand and Moslehifard (25) evaluated the tensile strength of dental acrylic resins using TiO₂ nanoparticles. They demonstrated that a significant increase was observed in tensile strength

using TiO₂ nanoparticle reinforcement agents up to 1% w/w; however, the opposite was noticed when a greater increase in TiO₂ nanoparticles was applied. The difference in the results may be attributed to different test methods and dental materials.

The measured hardness for Beautifil Flow increased by about 1.11 times compared to the control group by adding 1% w/w of TiO₂ nanoparticles. Nevertheless, the measured hardness the Clearfil AP-X Flow decreased to about one-third of the control group. However, the hardness of Clearfil AP-X Flow in all concentrations was higher than that of other flowable composites. In this context, a study was conducted by Yang Xia et al. in 2008 in which a group of samples was produced with TiO₂ nanoparticles whose surface was coated with a binding agent, and another group produced without a binding agent. They found that the specimens produced from TiO₂-coated with the binding agent increased in hardness (12). One of the possible reasons for obtaining better hardness in Beautifil Flow is related to its lower filler content. This lower concentration allows TiO₂ nanoparticles to be distributed in a more homogeneous mixture pattern. The filler content of Clearfil AP-X Flow was about 81% and as a result, inadequate space was available to distribute a homogeneous mixture of TiO₂ nanoparticles. Also, the type of fillers in each composite might affect the results. Clearfil AP-X Flow composite contains silanated fillers. Optimized silane content provided composites with the highest resistance against hydrolytic degradation. The silane treatment of fillers generally improved the flexural strength of the composites. The silane treatment of fillers improves the physical properties of the resultant composites and their resistance to hydrolytic degradation.

Moreover, this *in vitro* implementation requires long-term *in vitro* and *in vivo* studies with larger sample sizes, using various resin composites, adhesive systems, and different depths of dentin or different cavity preparations, since these factors may affect dentin bonding.

Conclusion

Due to the limitations of the present study, adding TiO₂ nanoparticles at very low concentrations up to 0.5% w/w positively affects the flexural strength of both Beautifil Flow and Clearfil AP-X Flow dental flowable composites. The hardness of Beautifil Flow with low or moderate filler content increased by adding TiO₂ nanoparticles; however, the opposite results were observed in Clearfil AP-X Flow.

Acknowledgments

The present article is extracted from the thesis of Dr. Sajjad Khaksabz. The authors would like to express their sincere gratitude to the Vice-Chancellor of Shiraz University of Medical Sciences for supporting the present study (Grant# 13513) and also to Dr. Salehi from the Dental Research Development Center for statistical analysis.

Conflict of Interest

The authors declare that there is no conflict of interest regarding the publication of the present study.

References

1. Baroudi K, Rodrigues JC. Flowable resin composites: A systematic review and clinical considerations. *J Clin Diagn Res.* 2015;9(6): 18-24.
2. Murchison DF, Charlton DG, Moore WS. Comparative radiopacity of flowable resin composites. *Quintessence Int.* 1999;30(3):179-84.
3. Clelland NL, Pagnotto MP, Kerby RE, Seghi RR. Relative wear of flowable and highly filled composite. *J Prosthet Dent.* 2005;93(2):153-157.
4. Gömeç Y, Dörter C, Dabanoglu A, Koray F. Effect of resin-based material combination on the compressive and the flexural strength. *J Oral Rehabil.* 2005;32(2):122-127.
5. Bonilla ED, Yashar M, Caputo AA. Fracture toughness of nine flowable resin composites. *J Prosthet Dent.* 2003;89(3):261-267.
6. Papia E, Arnoldsson P, Baudinova A, Jimbo R, VON STEYERN PV. Cast, milled and EBM-manufactured titanium, differences in porcelain shear bond strength. *Dent Mater J.* 2018;37(2):214-221.
7. Chung C-M, Kim J-G, Kim M-S, Kim K-M, Kim K-N. Development of a new photocurable composite resin with reduced curing shrinkage. *Dent Mater.* 2002;18(2):174-178.
8. Krause WR, Park SH, Straup RA. Mechanical properties of BIS-GMA resin short glass fiber composites. *J Biomed Mater Res A.* 1989;23(10):1195-1211.
9. Bayne S. Mechanical property analysis of two admixed PRIMM-modified commercial dental composites. *Acad Dent Mater Trans.* 1996;9:238.

10. Ghods K, Akbari P, Dehghani Ashkezari P, Ghayoomi S. Newest Developments of Nanotechnology in Dentistry: A Review of Literature. *J Dent Mater Tech.* 2022;11(1):1-10.
11. Fatemeh K, Mohammad Javad M, Samaneh K. The effect of silver nanoparticles on composite shear bond strength to dentin with different adhesion protocols. *J Appl Oral Sci.* 2017;25(4):367-373.
12. Xia Y, Zhang F, Xie H, Gu N. Nanoparticle-reinforced resin-based dental composites. *J Dent.* 2008;36(6):450-455.
13. Klapdohr S, Moszner N. New inorganic components for dental filling composites. *Monatsh Chem.* 2005;136(1):21-45.
14. Ibrahim MA, Meera Priyadarshini B, Neo J, Fawzy AS. Characterization of Chitosan/TiO₂ Nano-Powder Modified Glass-Ionomer Cement for Restorative Dental Applications. *J Esthet Dent.* 2017;29(2):146-156.
15. Sodagar A, Bahador A, Khalil S, Shahroudi AS, Kassaei MZ. The effect of TiO₂ and SiO₂ nanoparticles on flexural strength of poly (methyl methacrylate) acrylic resins. *J Prosthodont Res.* 2013;57(1):15-19.
16. Priyadarsini S, Mukherjee S, Mishra M. Nanoparticles used in dentistry; a review. *J Oral Biol Craniofacial Res.* 2018;8(1): 58–67.
17. Koohpeima F, Mokhtari MJ, Doozandeh M, Jowkar Z, Yazdanshenas F. Comparison of Cytotoxicity of New Nanohybrid Composite, Giomer, Glass Ionomer and Silver Reinforced Glass Ionomer using Human Gingival Fibroblast Cell Line. *J Clin Pediatr Dent.* 2017;41(5):368-373.
18. Uno M, Kurachi M, Wakamatsu N, Doi Y. Effects of adding silver nanoparticles on the toughening of dental porcelain. *J Prosthet Dent.* 2013;109(4):241-247.
19. Baloš S, Pilić B, Petronijević B, Marković D, Mirković S, Šarčev I. Improving mechanical properties of flowable dental composite resin by adding silica nanoparticles. *Vojnosanit Pregl.* 2013;70(5):477-483.
20. Teymoornezhad K, Alaghehmand H, Daryakenari G, Khafri S, Tabari M. Evaluating the Microshear Bond Strength and Microleakage of Flowable Composites Containing Zinc Oxide Nanoparticles. *Electron Physician.* 2016;8(11):3289–3295.
21. Sakaguchi R, Cross M, Douglas W. A simple model of crack propagation in dental restorations. *Dent Mater.* 1992;8(2):131-136.
22. Condon JR, Ferracane JL. Assessing the effect of composite formulation on polymerization stress. *J Am Dent Assoc.* 2000;131(4):497-503.
23. De Cesero L, de Oliveira EMN, Junior LHB, Papaléo RM, Mota EG. The addition of silica nanoparticles on the mechanical properties of dental stone. *J Prosthet Dent.* 2017;118(4):535-539.
24. Ziental D, Czarczynska-Goslinska B, Mlynarczyk DT, Glowacka-Sobotta A, Stanisiz B, Goslinski T, et al. Titanium dioxide nanoparticles: prospects and applications in medicine. *Nanomaterials.* 2020;10(2):387-418.
25. Shirkavand S, Moslehifard E. Effect of TiO₂ nanoparticles on tensile strength of dental acrylic resins. *J Dent Res Dent Clin Dent Prospects.* 2014;8(4):197-203.

Corresponding Author

Mohammad Javad Mokhtari

Department of Biology, Zarghan Branch, Islamic Azad University, Zarghan, Iran

Tel: +98713266911

Email: mj.mokhtari@iau.ac.ir